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L- or D-lactide prepn. - by heating L- or D-polylactic acid in presence of

a tin catalyst

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Patent

Application

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EP 261572 A 19880330 EP 1987113529 A 19870916 198813 B

BR 198704814 A 19880517 198824 E

JP 63101378 A 19880506 JP 1987234717 A 19870918 198824 E

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## Alerting Abstract EP A

Prodn. of L-(-)- and/or D-(+)-lactide comprises heating L-(-)- or D-(+)-polylactic acid (mean Mr about 400-2,000) in the presence of a catalyst (0.05-1.0wt.%), at 130-230 deg.C under reduced pressure, distilling off the lactide as it is formed. The catalyst is powdered tin, a tin halide, or an organic Sn cpd., pref. Sn(II) 2-20C alkanoates, alkenoates, hydroxyalkanoates, naphthoates or benzoates. The process is operated batchwise or continuously. Polylactic acid is obtd. by heating lactic acid with the above catalysts at about 150-170 deg.C, at reduced pressures.

ADVANTAGE - The process gives improved yields (e.g. 65-80%) of the optical enantiomorphs (99% purity).

## Equivalent Alerting Abstract US A

The (semi)continuous process for prepn. of L(-) or D(+)-lactide of substantial optical purity, comprises L(-)-polylactic acid as the starting material to make L(-) lactide and D(+)polylactic acid is used as the starting material to make D(+)-lactide, where the starting materials are of 90% optical purity, and where the polylactic acid is heated to 130-230 deg.C., under reduced pressure, in the presence of 0.05-1.0 wt.% of a

catalyst from tin dust, tin halide and organic tin cpds, derived from 1-20C carboxylic acids. Pref. the tin cpd. is tin lactate, tin tantrate, tin oxalate, tin dicaprylate, tin dilaurate, tin dipalmitate, tin distearate, tin dioleate, tin alpha-naphthoate, tin beta-naphthoate and tin dioctoate.

USE - Optically pure lactides are prepd. by the (semi)continuous industrial process.

USE - (4pp)